

STaRT-2012 SAMPLE TEST PAPER CLASS-XI (SC.-BIOLOGY)

Time : 90 min.

Maximum Marks : 200

GENERAL INSTRUCTIONS

- 1. he question paper contains 50 questions, 15 questions from Physics (1-15), 10 questions from Chemistry (16-25), 15 Questions from Biology (26-40) and 10 questions from Mental Ability (41-50).
- 2. The OMR sheet given in the examination hall is the Answer Sheet.
- 3. Blank papers, clip boards, log tables, slide rule, calculators, mobile or any other electronic gadgets in any form is not allowed.
- 4. Do not forget to mention your roll number neatly and clearly in the blank space provided in the answer sheet.
- 5. Each Question carries 4 marks. '1' mark will be deduct for each wrong answer. So attempt each question carefully.
- 6. No rough sheets will be provided by the invigilators. All the rough work is to be done in the blank space provided in the question paper.
- 7. In case of any dispute, the answer filled in the OMR sheet available with the institute shall be final.

Name : _____

Roll No. : _____

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A block of mass 2 kg is given an initial velocity of 10 m/sec. on a rough horizontal ground. Due to the friction, the block comes to the rest after travelling 5m distance. What should be the constant friction force acting on the block?

(A) 10 N (B) 20 N

(C) 30 N

- 2. In the circuit shown, reading of the ammeter and the voltmeter are respectively :
 - (A) 0.1 amp., 3 volt
 (B) 0.1 amp., 1 volt
 (C) 0.3 amp., 3 volt
 (D) 0.3 amp., 1 volt



(D) 40 N

- A 100 gm bullet is fired in forward direction with a velocity of 100 m/sec. Due to this, the free gun of mass 2 kg (excluding bullet) rebounds in backward direction. Total kinetic energy produced in this process is :
 (A) 500 J
 (B) 25 J
 (C) 525 J
 (D) 475 J
- 4. The graph of pressure excess v/s distance (x) for a sound wave is shown above. If the speed of sound is 300 m/sec, the time period of the vibration in the sound wave is : Pressure excess



- A candle is used as an object and placed at a distance of 30 cm from a lens of power 5D (power may be positive or negative). At how much distance from the lens, should we place a screen, so that a sharp and inverted image of the candle can be formed :

 (A) 30 cm
 (B) 40 cm
 (C) 50 cm
 (D) 60 cm
- 6. A motorcycle is moving along the path shown with constant speed 60 km/h. It takes two minutes to move from point P to point Q. The total length of the path from P to Q will be :

(A) 2 km

(B) 3 km

(C) 4 km

(D) 5 km









7.	Suppose a positive charge is moving with a velocity \vec{v} in a magnetic									
	field \vec{B} , and experiences a magnetic force \vec{F} . According to the Fleming's									
	left hand rule, the fore-finger, the central finger and the thumb will									
	respectively point towal	rds :								
	(A) B, V and F		(B) V,Band F							
	(C) \vec{F}, \vec{V} and \vec{B}		(D) None of these							
8.	Which of the following	objects will float in water :								
	(A) mass = 50 g and vo (C) mass = 40 g and vo	$plume = 20 \text{ cm}^3$	(B) mass = 200 g and volume = 500 cm^3							
	(C) mass = 40 g and ve									
9.	A cubical block is proj displacement 'x' when i be :	ected horizontally on a retrieved to	rough surface with speed v and it stops sliding afte 2v, displacement of the block before it stops sliding will							
	(A) x	(B) 2x	(C) 4x	(D) 0.5 x						
10.	An electron enters a ma	gnetic field at right angles	to it, as shown in figure.	The direction of force acting on						
	the electron will be :									
			Magnatic field							
		Electron								
	(A) to the right	(B) to the left	(C) out of the page	(D) into the page						
11.	Which of the following lenses would you prefer to use while reading small letters found in a dictionary.									
	(A) A convex lens of for	the dictionary ?	(B) A concave lens of f	ocal length 40 cm						
	(C) A convex lens of foc	cal length 5 cm	(D) A concave lens of focal length 5 cm							
12.	Equivalent resistance b	etween A and B is :		D						
			(R) R							
	(A) 2K		(B) 2 Ao-	NNR B						
	(C) $\frac{R}{3}$		(D) $\frac{3R}{2}$	R						
13.	Two particles P and Q	start from rest and mov	e for equal time on a s	traight line. Particle P has an						
	acceleration of X m/s ² for the first half of the total time and 2X m/s ² for the second half. The particle Ω has an acceleration of 2X m/s ² for the first half of the total time and X m/s ² for the second half.									
	Which particle has cov	vered larger distance?								
	(A) both have covered	the same distance	(B) P has covered the larger distance (D) none of these							
	(C) Q has covered the larger distance (D) none of these									







14.	 The minimum work done to accelerate a block on a smooth horizontal surface from rest to speed v (A) is less than that required to accelerate if from v to 2v. (B) is equal than that required to accelerate it from v to 2v. (C) is more than that required to accelerate it from v to 2v. (D) may be any one of the above since it depends on the force acting on the truck and the distance over which it acts. 								
15.	Three resistance of value 1 Ω , 2 Ω and 3 Ω are connected in parallel. If the effective resistance of the circuit has to be 1 Ω , the value of the resistance to be connected in series to this circuit should be :								
	(A) <u>6</u> 11Ω	(B) ⁵ / ₁₁ Ω	(C) $\frac{4}{11}\Omega$	(D) $\frac{3}{11}\Omega$					
16.	Which of following is res (A) $Ca(HCO_3)_2$	sponsible for temporary h (B) Na ₂ CO ₃	ardness of water ? (C) CaCO ₃	(D) MgSO ₄					
17.	A solution turns red litm (A) 1	nus blue, its pH is likely to (B) 4	o be : (C) 7	(D) 10					
18.	An atom which has a m (A) isomer of nitrogen	ass number of 15 and ha (B) isobar of oxygen	as 7 neutrons is an : (C) isotope of oxygen	(D) isobar of carbon					
19.	10 mL of a solution of NaOH is found to be completely neutralised by 8 mL of a given solution of HCl. If we take 20 mL of the same solution of NaOH, the amount of H_2SO_4 solution (having the same molarity as that of initial HCl solution) required to neutralise it will be :								
20.	 Choose the correct statement with regard to redox displacement reactions. (A) A less active metal displaces a more active metal. (B) A more active non-metal is displaced by a less active non metal. (C) A less active non-metal displaces hydrogen from dilute acids. (D) A more active metal displaces hydrogen from dilute acids. 								
21.	Element X forms a chloride with the formula XCI_2 , which is a solid with a high melting point. X would most likely be in the same group of the Periodic Table as ? (A) Na (B) Ca (C) Al (D) Si								
22.	 Which of the following statements is not a correct statement about the trends when going from left to right across the periods of peridic Table - (A) The atoms generally become smaller in size. (B) The number of valence electrons increases. (C) The atoms lose their electrons more easily. (D) The oxides become more acidic. 								
23.	Consider the following equilibrium situation and identify the correct statement : $2H_2(g) + CO(g) \rightleftharpoons CH_3OH(g)$								







	(A) Addition of H_2 (g) would lead to increased production of CH_3OH (g). (B) Addition of CH_3OH (g) would stimulate further consumption of CO (g). (C) Increasing pressure would lead to the production of H_2 (g). (D) Reducing the volume of the equilibrium will not disturb the equilibrium.									
24.	3 moles of an ideal gas of temperature of the gas i (A) 50L	occupices 100L under cert s doubled keeping the pro (B) 200L	ain conditions. 1.5 moles essure constant. What is (C) 400L	of the gas is removed and the the new volume of the gas ? (D) 100L						
25.	Decomposition reaction to 40% in 10 min starting (A) 6×10^{-3}	of ammonia is given as 21 g with 0.1mol/L. The aver (B) 4 × 10 ⁻³	$NH_3 \longrightarrow N_2 + 3H_2$. The concentration of NH_3 decreat rage rate of decomposition of NH_3 in mol L ⁻¹ min ⁻¹ is (C) 4 × 10 ⁻² (D) 6 × 10 ⁻²							
26.	Archaebacteria differ fro (A) Mucopeptide substa (C) Prokaryotic nature	m true bacteria in not hav nces in cell wall	ing (B) Incipient nucleus (D) Cell organelles	PE						
27.	The bacterial genera carrying out nitritication, nitration, asymbiotic and symbiotic nitrogen fixation are (A) Rhizobium, Azotobacter, Nitrosomonas, Nitrobacter (B) Nitrosomonas, Nitrobacter, Azotobacter, Rhizobium (C) Nitrosomonas, Nitrobcater, Azotobacter, Clostridium (D) Nitrosomonas, Nitrobacter, Azotobacter, Rhizobium									
28.	Nitrogenase occurs in se (A) Akinites	ome blue-green algae in i (B) Heterocysts	ts (C) Hormogones	(D) Endospores						
29.	Red tide is caused by (A) Ceratium	(B) Noctiluca	(C) Gonyaulax	(D) All of these						
30.	Deuteromycetes are cal (A) True sexuality	led fungi imperfecti as the (B) Perfect stage	ey lack (C) Fruiting body	(D) All of these						
31.	Xylem with vessels is for (A) Cycas and Pinus (C) Araucaria and Taxus	und in which gymnospern	rms (B) Ephedra and Gnetum (D) Thuja and year							
32.	Synthesis of ATP in mito (A) In matrix (C) In intracristal space	ochondria takes place	(B) Upon cristae (D) At outer membrane							
33.	Raphides are crystals of (A) Calcium oxalate (C) Calcium phosphate	f	(B) Calcium carbonate (D) Magnesium carbonate							







34.	Species are considered as- (A) Artificial concept of human mind which cannot be defined in absolute terms (B) Real units of classification devised by taxonomists (C) Real basic units of classification (D) The lowest units of classification											
35.	In Amoeba and paramecium osmoregulation occurs through-											
	(A) Pseudopodia	(B) Contractile vacuole	(C) Nucleus	(D) General surface								
36.	What happens if a bone	of frog is kept in dilute h	ydrochloric acid ?									
	(A) Shrink in size	(B) turn flexible	(C) Crack into pieces	(D) Assume black colour								
37.	An organic substance b	ound to an enzyme and e	essential for its activity is	called-								
	(A) Isoenzyme	(B) Coenzyme	(C) Holoenzyme	(D) Apoenzyme								
38.	Male and female genital pores in pheretima are found on which segment :(A) 14th & 18th segment(B) 18th & 14th segment(C) Inter-segmental grovves of 5/6(D) Inter-segmental grooves of 6/7											
39.	National animal of India	is-										
	(A) Panthera leo	(B) Panthera tigris	(C) Acinonyx	(D) panthera pardus								
40.	In relation to digestion which one of the following is the correct matching of the site of action, given substrate, the enzyme acting upon it and the end product -											
	(A) Buccal cavity - Bact	eria <u>Lysozyme</u> A.A a	nd F.A									
	(B) Duodenum - Lipid $\xrightarrow{\text{Lipase}}$ F.A. and glycerol											
	(C) Stomach - caseinPepsin Calsium paracasinate											
	(D) Stomach of ruminent - cellulose $\xrightarrow{\beta-amylase}$ D-glucose											
Directio	ons : (41 to 43) Find the	e missing term :										
	(A) 210	151, 158, 172, 182, ? (A) 210 (B) 193 (C) 197 (D) 203										



Directions : (44 to 45) Column I contains five capital letters while column II contains five digits. Each letter corresponds to a single digit but not necessarily in that order. TPAP

स्तम्भ (column) l	स्तम्भ (column) II
BEIKL	61520
PNBTK	34568
XLPBE	57401
KNIXV	27396
XBNPE	45713

- 44. What is the value of BIKE ? (A) 5261 (B) 6125 (C) 2560 (D) None of these 45.
- What is the value of PIN + NIP ? (A) 423 (B) 744 (C) 777 (D) 747
- 46. In a row of girls, Rina and Mona occupy the ninth place from the right end and tenth place from the left end, respectively. If they interchanged their places, Rina and Mona occupy seventeenth place from the right and eighteenth place from the left, respectively. How many girls are there in the row ? (B) 26 (C) 27 (D) Data inadequate (A) 25
- 47. At what time between 4 and 5 will be hands of clock be in opposite direction ?

(A)
$$53\frac{7}{11}$$
 min. past 4 (B) $21\frac{9}{11}$ min. past 4 (C) $54\frac{6}{11}$ min. past 4 (D) $49\frac{1}{11}$ min. past 4







Directions : (48 to 49) Each of the following questions consists of five figures marked 1, 2, 3, 4 and 5. These figures form a series. Find out the one from the answer figures that will continue the series. Problem Figures



50. On the basis of the following figures you have to tell which number will come in place of '?'

6 1	4 6	7 5		
(i)	(ii)	(iii)		
(A) 2	(B) 3	3	(C) 6	(D) 4

		A		SWE	R										
1.	(B)	2.	(B)	3.	(C)	4.	(C)	5.	(D)	6.	(A)	7.	(A)	8.	(B)
9.	(C)	10.	(D)	11.	(C)	12.	(C)	13.	(C)	14.	(A)	15.	(B)	16.	(A)
17.	(D)	18.	(C)	19.	(B)	20.	(D)	21.	(B)	22.	(C)	23.	(A)	24.	(D)
25.	(A)	26.	(A)	27.	(B)	28.	(B)	29.	(C)	30.	(D)	31.	(B)	32.	(B)
33.	(A)	34.	(C)	35.	(C)	36.	(B)	37.	(B)	38.	(B)	39.	(B)	40.	(B)
41.	(B)	42.	(D)	43.	(B)	44.	(A)	45.	(D)	46.	(B)	47.	(C)	48.	(D)
49.	(D)	50.	(B)												